

The Influence of Tin Oxide Doping on Structural and Optical Properties of (Bi₂O₃) Thin Films

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Abstract

In this research, the powder mixing technology has been used to control the ratio of added, where the addition of tin oxide to the bismuth oxide of by (1,3,5,7)% by using vacuum evaporation method. The range of thickness for thin films (40-100)nm . EDX analysis shows the results of the ratios were close to the required values. XRD analysis shows pure bismuth oxide crystalline of the tetragonal phase (β -Bi₂O₃), observed that the structural properties improves with addition (1)% and the composition of a triangular compound (Bi₆₄SnO₉₈), and with addition (5)% we have a new triple compound (Bi₁₂SnO₂₀). By increasing the impurities, the structural properties become close from random, the tin oxide begins with bismuth oxide separation. AFM measurement shows the thin films surface a pekes like (nano needle) and the grain size rises to (20.2-22 nm). In the case of increased impurities, the surface becomes smooth, and the nano needle decrease with grain size . The optical properties showed a rise in the value of the optical power gap (2.6-3)eV and directly allowed transitions are simply increasing from (3.3-3.5) eV, and directly forbidden transitions are simply increasing to from (2.48-2.73) eV.

Key words: .Bismuth oxide . Structure Properties (Bi₂O₃). Optical properties(Bi₂O₃)

1. Introduction

Bismuth oxide Bi₂O₃ is one of the most important semiconductor substances that researchers are interested in studying [1] because of its optical and electrical properties such as large energy band gap, [2]. So far it is known that Bismuth has five main polymorphic forms α - β - γ - δ and ω [3] Each of them has polymorph possesses different crystal structures and various energy band gap and mechanical properties [4]. α -phase stabilizes in monoclinic at low temperature and high temperature δ -phase centered cubic are stable, β -phase are tetragonal, γ -phase are body-centered and ω -phase are triclinic in high temperature [5]. Bi₂O₃ is prepared with various techniques including vacuum evaporation, chemical analysis , chemical bath deposition and pulsed laser deposition [6]. The properties of Bi₂O₃ are altered by doping impurities to it that increase the ability to control optical and structural properties , as well as the improvement in electrical conductivity [7].

2 . Experimental

2-1- maxing

In this research, tin and bismuth were used (Indian industry) in the form of high purity powder up to 99.9% . Bismuth was used as base material (1-X) and tin as an additive and in weight $x=(0.01,0.03,0.05,0.07)$ gram, Table(1) shows . After adding the tin to the bismuth, mix the mixture using a mortar of Agate for a period of (60 minutes) . Then placed in mold of steel and press to (5 tons/cm²) for (5 minutes) and then heated in the oven (German-made) to 100°C . The size of the capsule is (2.5x10)mm .

Table(1) the weight of additions to prepare samples.

Samples	Bi (g)	Sn (g)
Pure	1	0
1%	0.9921	0.0133
3%	0.9743	0.0384
5%	0.9739	0.0591
7%	0.9358	0.0771

2-2- Method and Deposition

The deposition was done using a vacuum evaporation system in vacuum (1.8 X 10⁻⁴mpa) experimental technique to determine the thickness of the thin film . The experiment showed that the weight of the object inside a molybdenum boat gives the thickness of a thin film, which is deposited on a glass base (10.5)cm height from the boat. The process evaporates relatively quickly so that the mixture cannot decompose because of the melting point difference. The molybdenum boat was chosen as the mixture does not stick to it and when evaporated it floats on the boat but sticks to the tungsten boat. Thin films were heated under low pressure 1x10⁻³ at 200 ° C for 60 minutes and left to cool before compressing pressure with atmospheric pressure. The thin film is oxidized in an oven with air temperature at 350 ° C for 60 minutes. The weighted method represented To calculate thickness, by equation (1)[8] .

$$t = \frac{m_2 - m_1}{A \cdot \rho_f} \text{ ----- (1)}$$

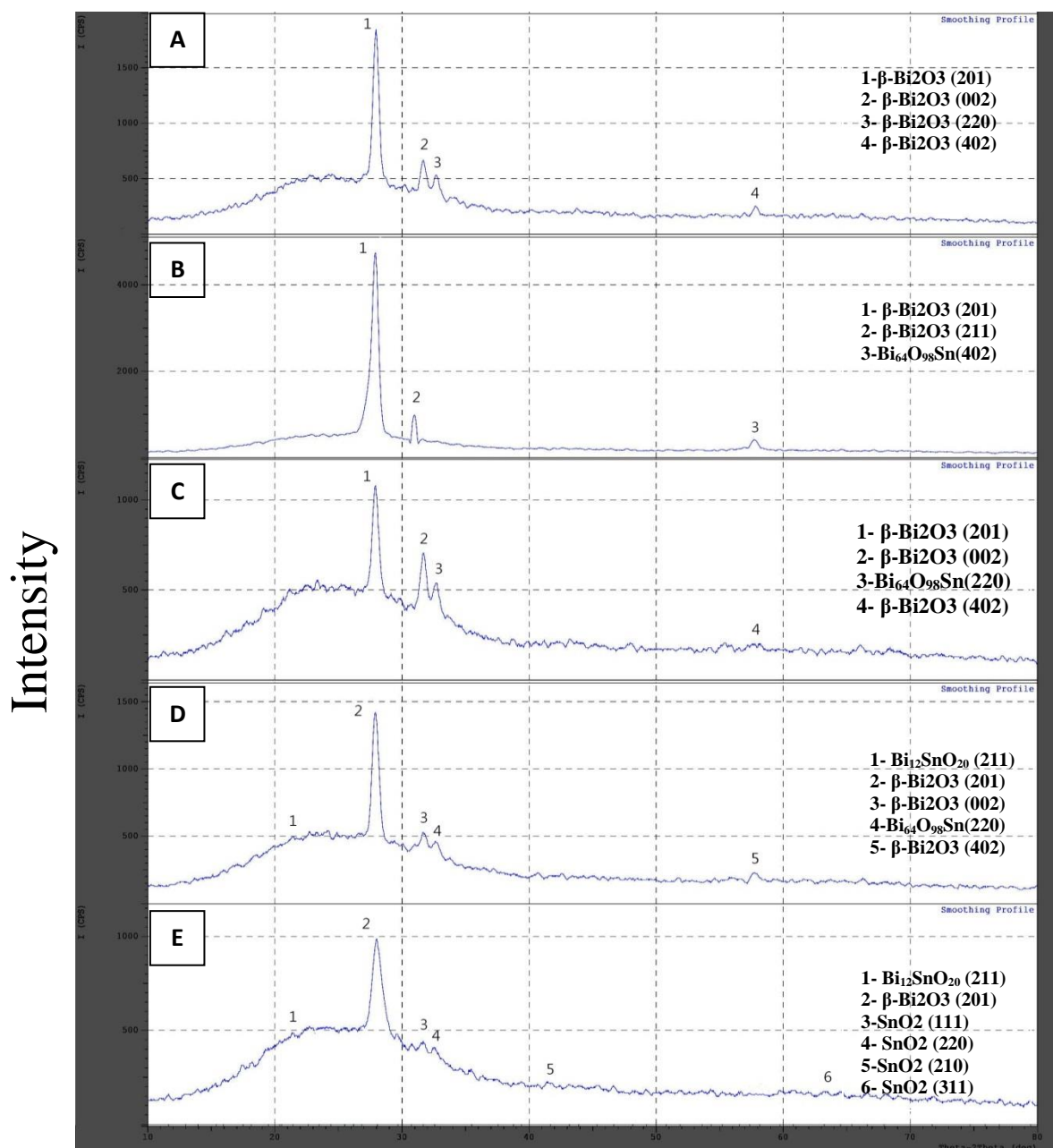
Where t: thickness, $m_2 - m_1$: difference weight between m_2 and m_1 , ρ_f : Intensity of material , A: The area of the thin films.

3. Results and discussion

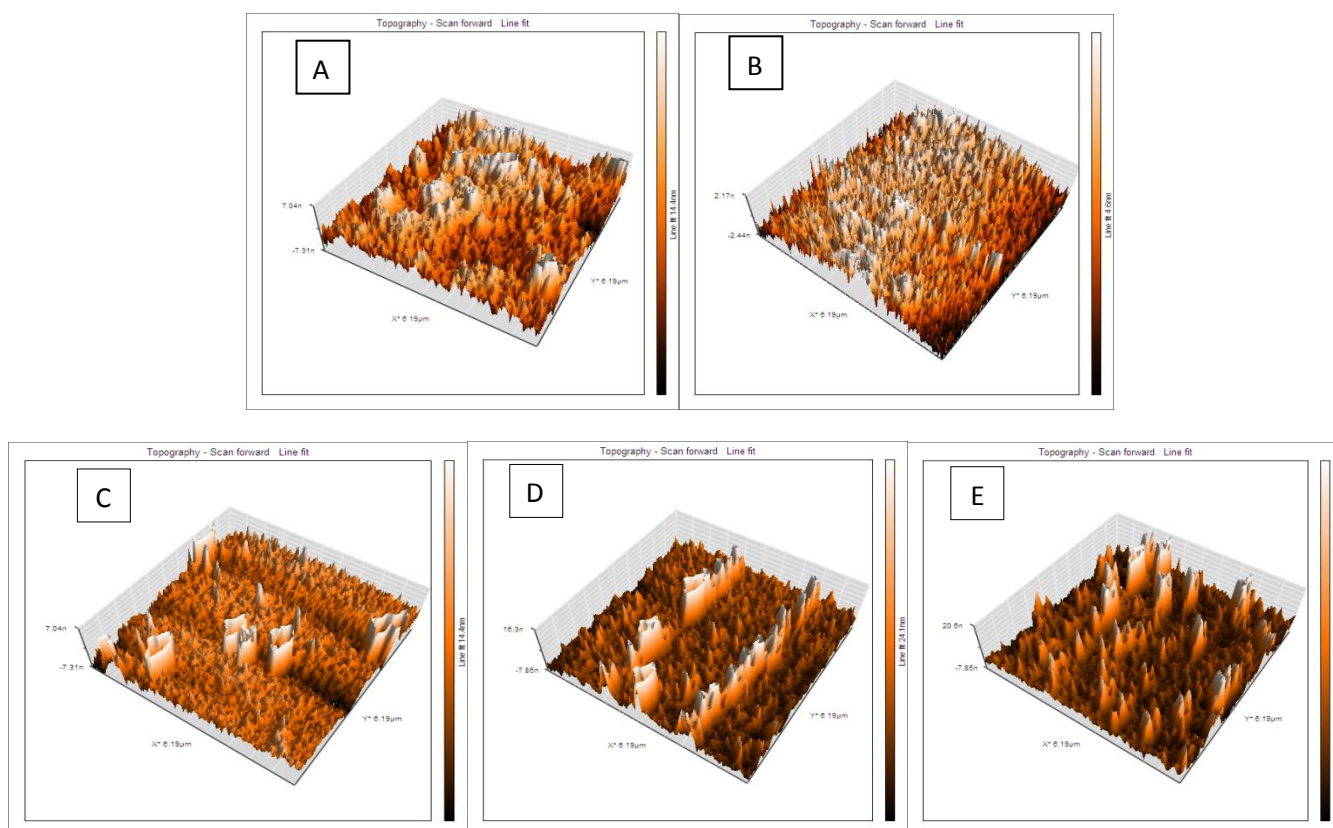
3-1- Structural properties

Figure (1) shows the results of the analysis structural properties of an different device type (SHIMADZU-Japan-XRD 6000). the thin film is composed of pure bismuth oxide (β -Bi₂O₃), the results shows that the thin film has a crystalline structure of the tetragonal phase in which the directivity (201) is predominant at the angle (27.9)

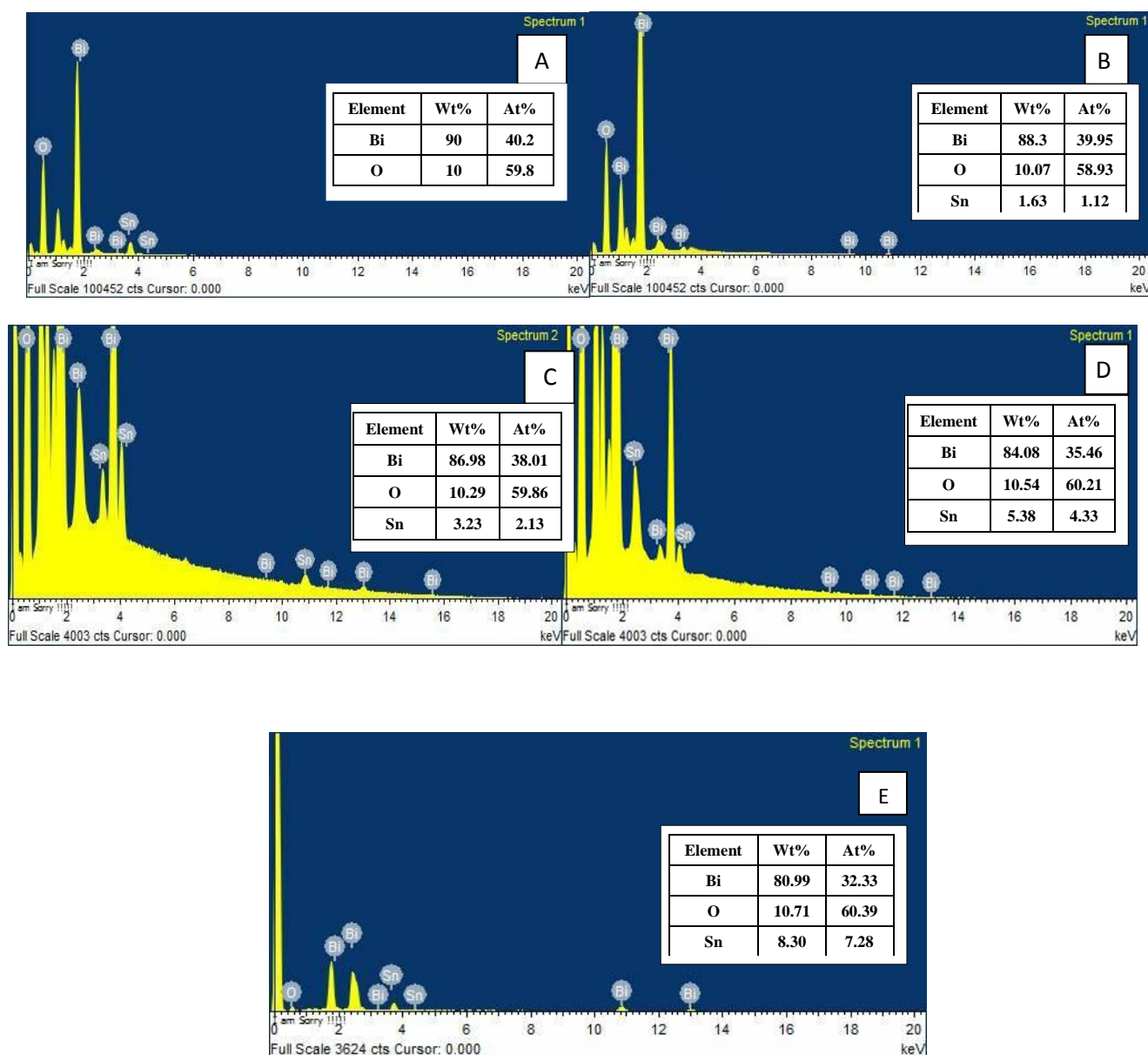
followed by the directivity (002) at the angle (31.7) ,followed by the directivity (220) and (402) in angles (32.7 and 57.7), these results were agreed with researchers [4,9,10,11,16]. The method of preparation produced a structure in the form of a nano needle up to (2.17) nm with grain size (20.2), topography by an AFM device (Nano PHYWE-German) show in Figure (2-A). The element analysis system (EDX) shows that the thin film composed of bismuth oxide, Bismuth weight of (90%) and (10%) relative to oxygen as shown in Fig (3-A). The results were agreed with international cards (American Standard For Testing Material) no: (027-0050) [16,11]. At the addition ratio ($x=0.01$) gram the thin film is composed ($\text{Bi}_2\text{O}_3(0.99)\text{SnO}_2(0.01)$). The results of the (x-Ray) analysis shows an improvement in the crystalline structure of the base compound with directivity remaining (201) at the angle (27.9) followed by directivity (211) at the angle (30.9) only, The analysis shows a triple compound ($\text{Bi}_{64}\text{SnO}_{98}$) at directivity (57.8)The results were agreed with (ASTM) cart no: (042-0192). Figure (2-B) show Topography by an AFM device for, shows the structural nano needle with columns of up to (7.04)nm with grain size (20.4)nm. The (EDX) shows the ratio of (Sn) is 1,63%, which is slightly higher than the target ratio[11] . By increasing the value of x to (0.03) gram the structural properties of the then film shows a decrease in the intensity value of the X-ray deflected and change the angular direction of the ($\text{Bi}_{64}\text{SnO}_{98}$) to angular (32.6) dy directivity (220) and (Bi_2O_3) remained in the angular (27.9) dy directivity (201) , with up the value of surface compositions to (20.6) nm with grain size (22 nm) show in Fig(2-C) .The (EDX) shows the ratio of (Sn) is 3,23% the results were agreed with [10,11,13-16] with different method and materials.



(Figure (1) XRD- A- Pure ; B- 1% ; C- 3% ; D- 5% ; E - 7%)



(Figure (2) AFM A- Pure ; B- 1% ; C- 3% ; D- 5% ; E- 7%)



(Figure (3) EDX - A- Pure ; B- 1% ; C- 3% ; D- 5% ; E- 7%)

At the arrival of the addition ratio at (X=0.05) gram the triple compound ($\text{Bi}_{64}\text{SnO}_{98}$) can be observed in the same angle with a decrease in the value of the intensity as an indicator, but the analysis shows a new chemical triple compound ($\text{Bi}_{12}\text{SnO}_{20}$) according to the international card (ATSM) the numbered (054-1004) angle(21.4) by the directivity (211) and the percentage of add impurities in the card for the tin oxide is (4.03)%. The topography of the surface is in Fig. (2-D), Shows a decrease in peaks to (7.5) with a decrease in the number of these peaks and a low grain size to (21.2)nm . The result of (EDX) analysis shows that the ratio of the tin oxide is equal to (5.38)%, with an increase in the value of X [10-13]. When increasing the addition to (X=0.07) gram ,the analysis shows a one peak of the bismuth oxide at the angle (27.9) with the triple compound ($\text{Bi}_{12}\text{SnO}_{20}$) at angle (21.4) and the disappearance of compound ($\text{Bi}_{64}\text{SnO}_{98}$), other peaks in these diffraction are for tin dioxide, this is due to the increase in the percentage of added impurities and that the

tin oxide has been separated from the bismuth oxide and tapped on the surface. This addition has affected the crystalline structure leading to the random direction and decrease in the intensity of X-ray and particle size. The analysis of (EDX) shows that the ratio of the addition is (8.30), has increased, this confirms that there are loss ratios in the base substance Bismuth.

3-2- Optical properties

The optical properties of the thin films on glass bases were studied using (Shimadzu1800-Japan spectrometer), where it measured permeability and absorption with a range of (300-900) nm . Figure (4) shows the optical properties of pure bismuth where the absorption coefficient calculated using formula (2) [16]

$$\alpha = 2.303(A/t) \dots \dots \dots (2)$$

Where , t:is thin film thickness , A :absorption . And energy gap calculated using formula(3) [16]

$$(h\nu\alpha)^y = \beta(h\nu - E_g) \dots \dots \dots (3)$$

Where $h\nu$ is energy of photon : β is a parameter : y is index take the value(2) to calculate direct allowed transitions and(2/3) calculated direct forbidden transitions . The results showed that the optical energy gap (2.58)ev, direct allowed transitions (3.3)ev, and the direct forbidden transitions (2.48)ev. These results were agreed with researchers [1,10,11,15,16] Figure (4) shows the optical properties of bismuth oxide thin film mixed with tin oxide . The results showed that the optical energy gap increases with increasing levels of tin oxide. Where it increased from (2.6ev to 3ev), excluding the deviation by (3%) where it decreased to (2.3ev), the results were similar to the researcher [10,11], which was used for the doping using aluminum ,cerium. Where tin oxide at the ratio of (3%) to the formation of new energy levels above the valance level. Increase the energy gap of the allowed direct transitions (3.3eV to 3.5eV) and Increase the energy gap of the forbidden direct transitions (2.48ev to 2.73ev). As a result of a reorganization in energy levels.

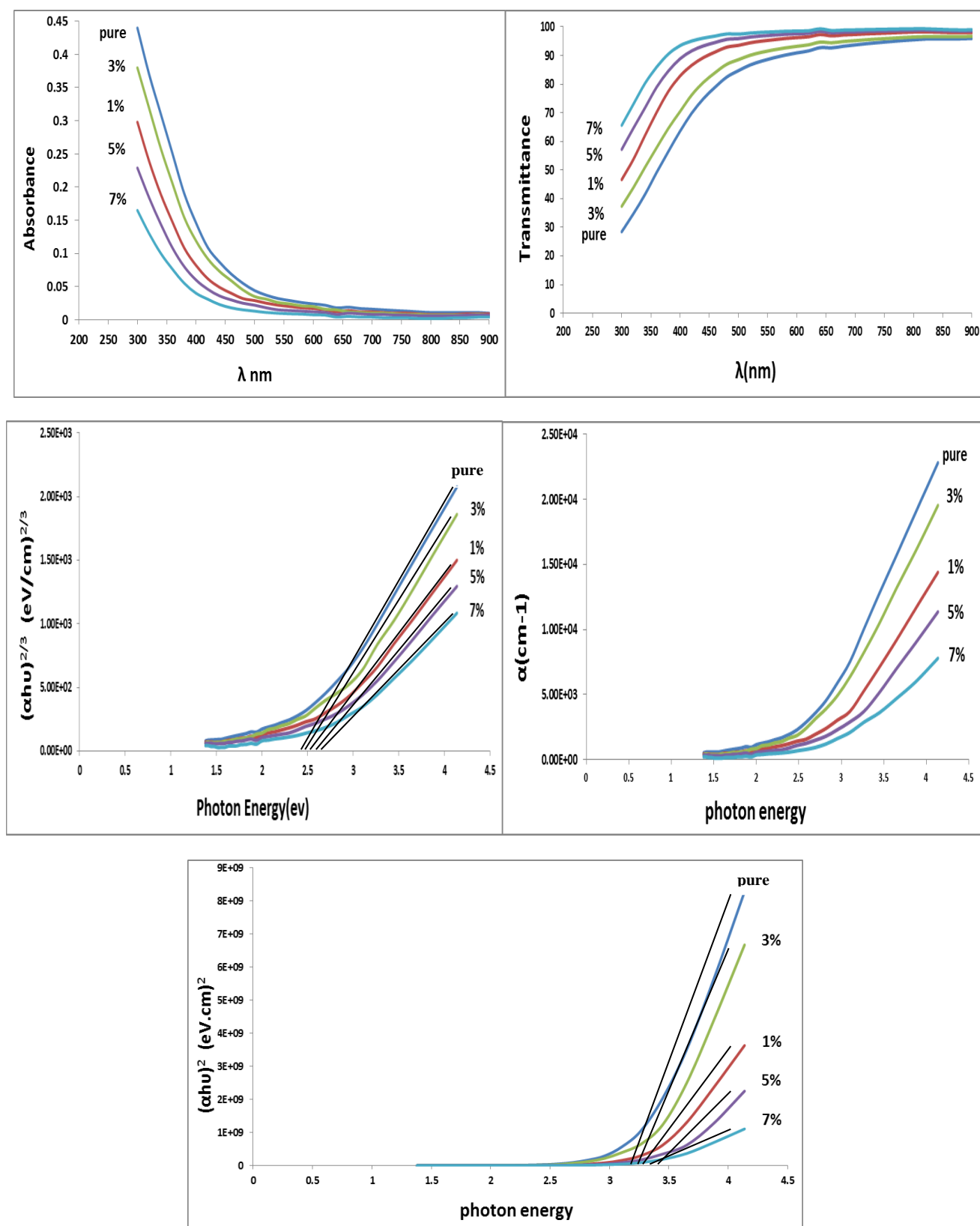


Figure (4) Results of UV-VIS

Conclusion

Bismuth Oxide, Prepared by pressing and vacuum evaporation deposited on the glass substrate has β -phase. Structural properties showed an improvement by adding tin oxide (1%) and obtaining the compound ($\text{Bi}_{64}\text{SnO}_{98}$), in addition (5%) we obtained a new compound ($\text{Bi}_{12}\text{SnO}_{20}$), this proves that the material can be composed. The addition of the tin oxide by mixing and the pressure method using the weight ratios was close to the required ratios. Increasing the additive ratio nearing from random structural properties with the separation of tin oxide from bismuth oxide. The addition of tin oxide increases the optical energy gap.

CONFLICT OF INTERESTS.

There are non-conflicts of interest.

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الخلاصة

في هذا البحث ، تم استخدام تقنية خلط المساحيق للتحكم في نسبة الإضافة حيث تمت إضافة أكسيد القصدير الى أكسيد البزموت بنسبة (1, 3, 5, 7) % باستخدام طريقة التبخير بالفراغ ، بمدى سمك للأغشية الرقيقة (40-100) نانومتر . اظهر تحليل مجهر الماسح الالكتروني ان نتائج النسب كان قريبة من القيم المطلوبة . واطهر تحليل الاشعة السينية ان أكسيد البزموت النقي البلوري من الطور الرباعي (β -Bi₂O₃) ، لوحظ ان الخصائص التركيبية تتحسن عند إضافة (1%) ونحصل على المركب ثلاثي (Bi₆₄SnO₉₈) ومع إضافة بنسبة (5%) نحصل على مركب ثلاثي جديد (Bi₁₂SnO₂₀) ، وبزيادة نسبة الشوائب تصبح الخصائص التركيبية قريبة من العشوائية ، ويبدأ أكسيد القصدير بالانفصال من أكسيد البزموت . اظهر مقياس AFM ان الاغشية الرقيقة تظهر على شكل قمع تشبه الابر النانوية ، وفي حالة زيادة الشوائب يصبح السطح ناعم ، أظهرت الخواص البصرية ارتفاعا في قيمة فجوة الطاقة الضوئية ev (2.6-3) فجوة الطاقة المباشرة ev (3.3-3.5) وفجوة الطاقة غير المباشرة تزيد نسبيا أيضا ev (2.48-2.73).

الكلمات الدالة: أكسيد البزموت ، الخواص التركيبية (Bi₂O₃) ، الخواص البصرية (Bi₂O₃)